

# Photometrische Methoden:

## UV/VIS Photometer

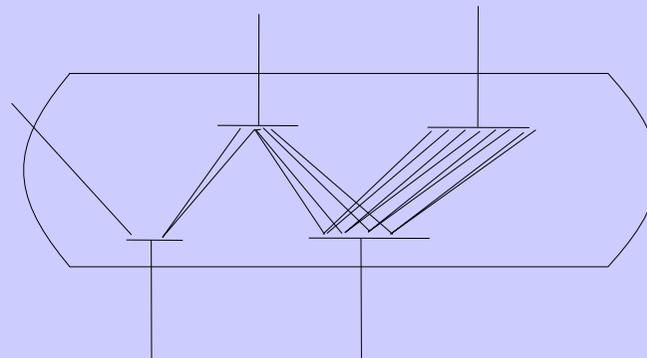
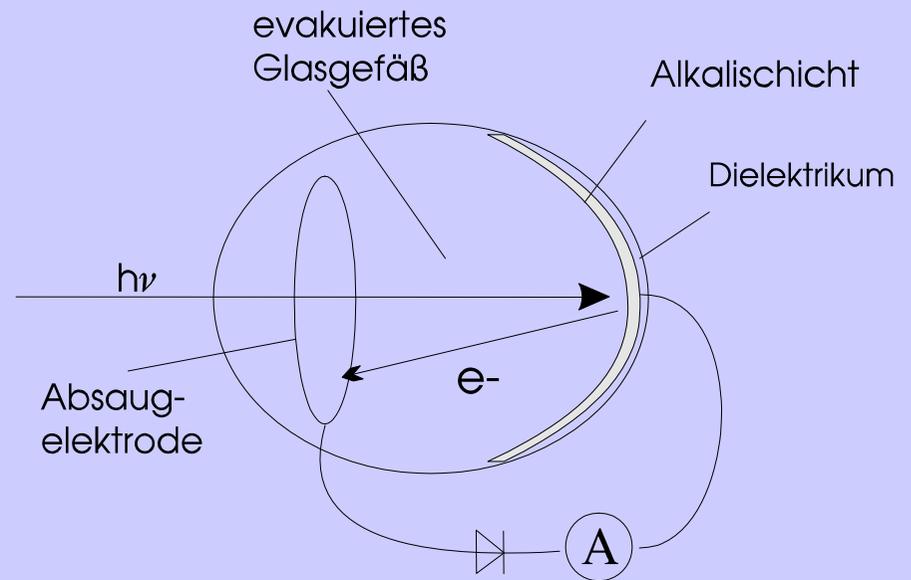
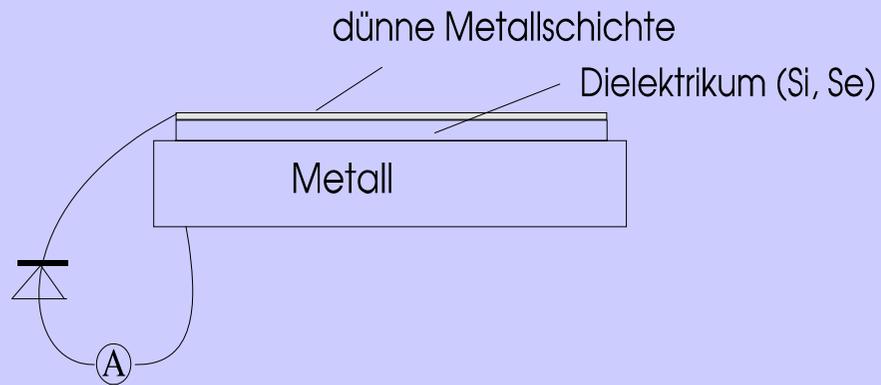
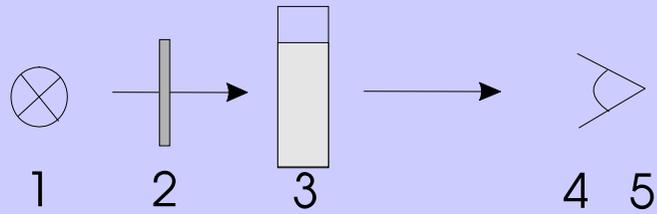
### Grundlagen 1:

- Lichtmessung
- spezifische Lichtabsorption, Monochromasie
- Zusammenhang: Konzentration und Lichtschwächung  
das Lambert- Beersche Gesetz

### Messpraxis:

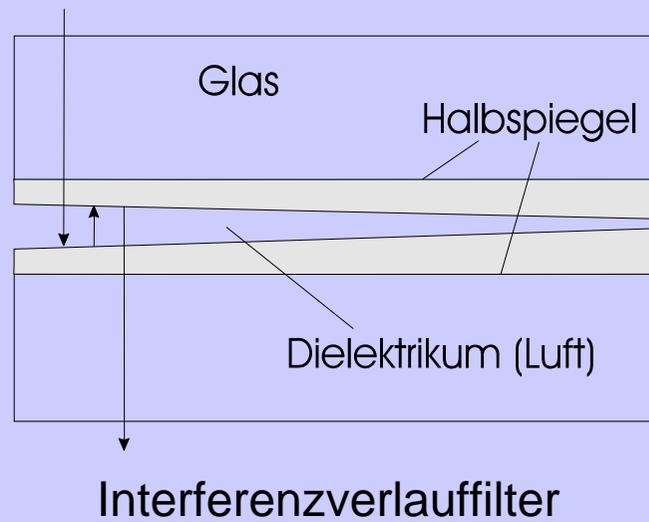
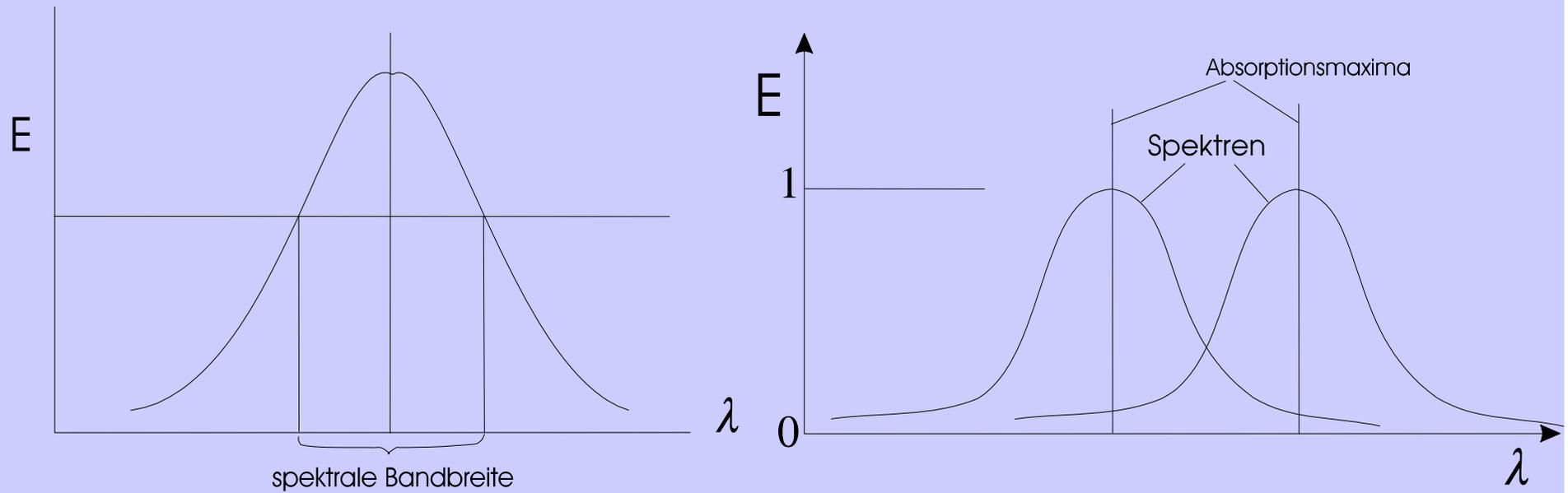
- Konzentrationsmessung (Mikroplatte), Kalibration
- Absorptionsspektrum (Diondenarray)
- Enzymkinetik (Zweistrahlfotometer)

# Lichtmessung

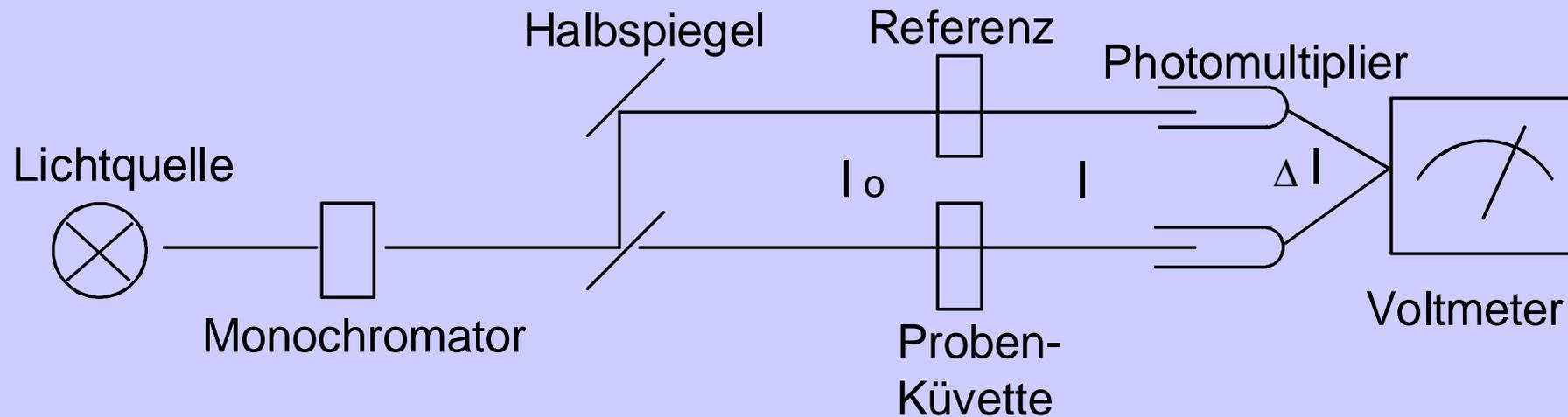


Sekundärelektronenvervielfacher - photomultiplier

# Monochromatisches Licht



# Photometrie



$I_0, I$  ... Intensität

$E$  ... Extinktion bzw.  $A$  ... Absorption

$\epsilon_\lambda$  ... molarer Extinktions- bzw. Absorptionskoeffizient

$d$  ... Schichtdicke der Küvette

$c$  ... Konzentration

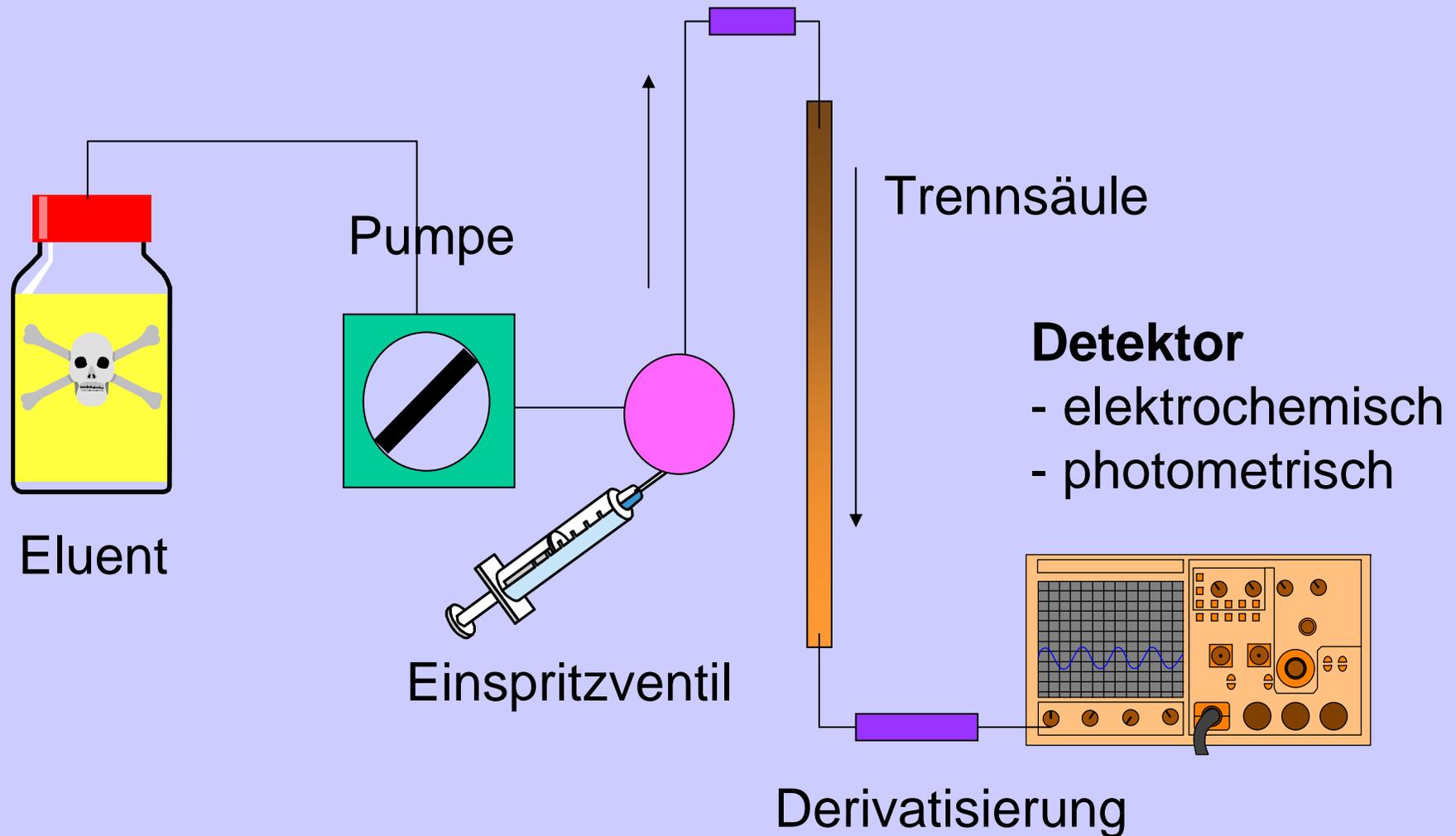
$$A = \lg \frac{I_0}{I} = \epsilon_\lambda c d \dots \text{Lambert und Beer}$$

$$c_{\text{Probe}} / a_{\text{Probe}} = c_{\text{Ref}} / a_{\text{Ref}}$$

Wiederholung

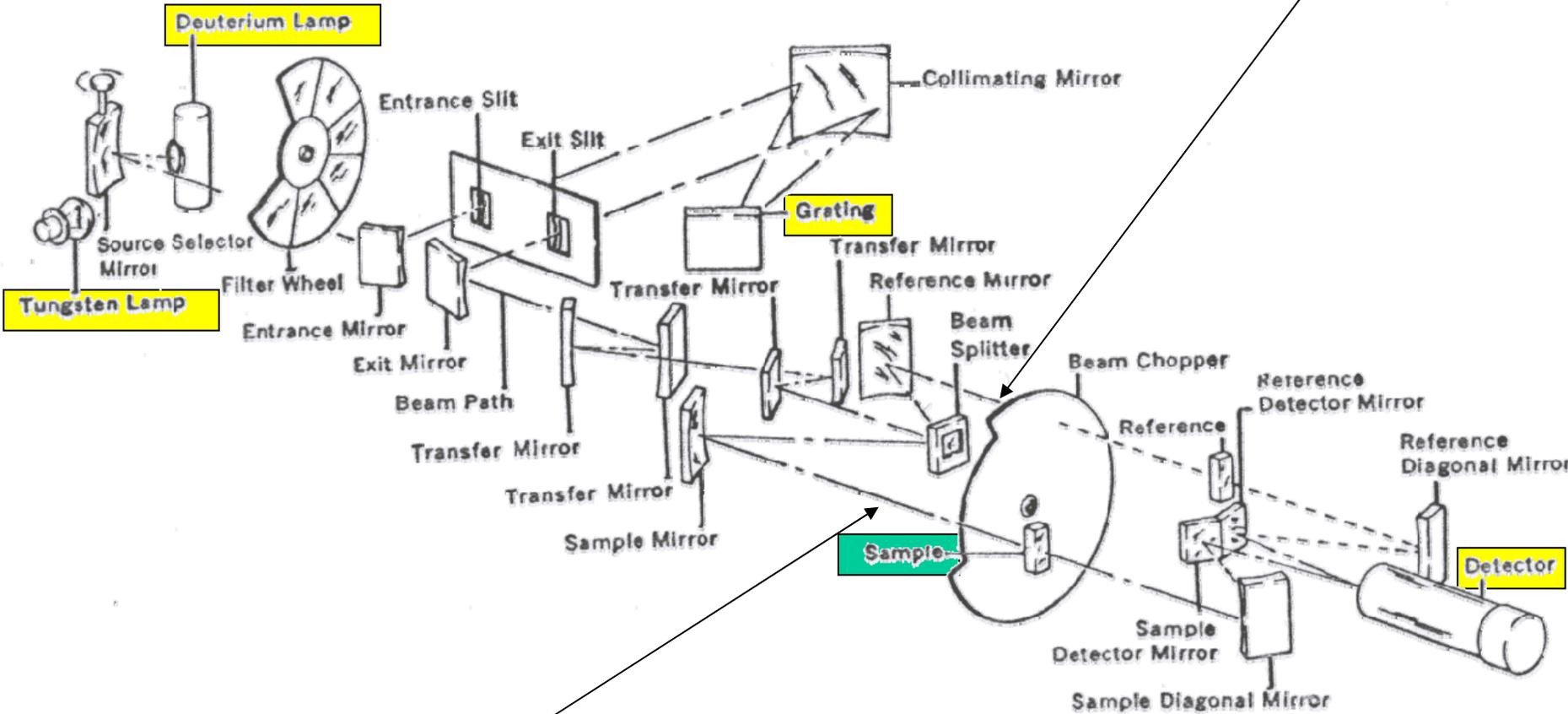
# Flüssigkeitschromatographie

Derivatisierung



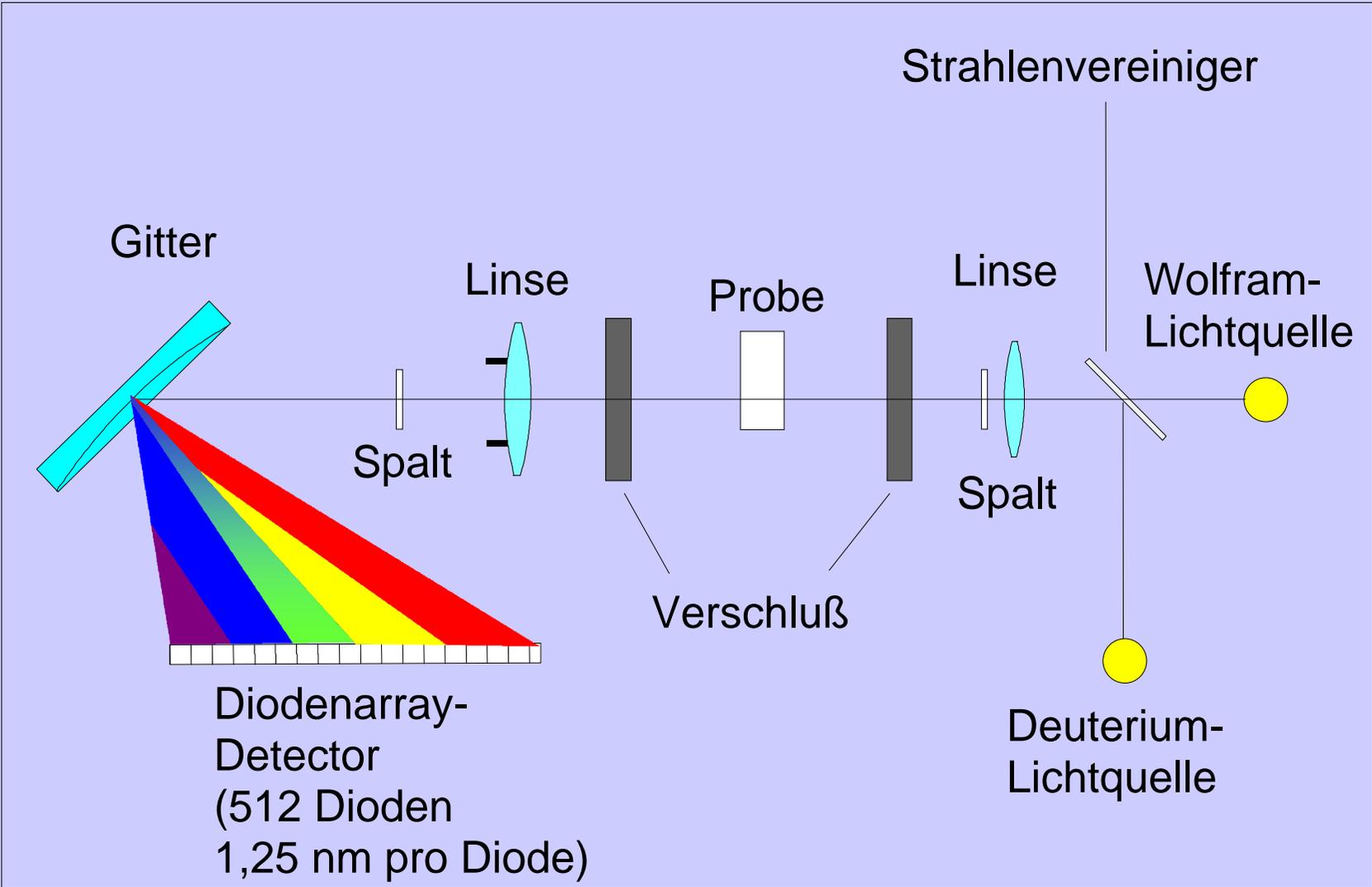
# Zweistrahlfotometer:

Referenzstrahlengang



Probenstrahlengang

# DIODENARRAY- SPEKTRALPHOTOMETER

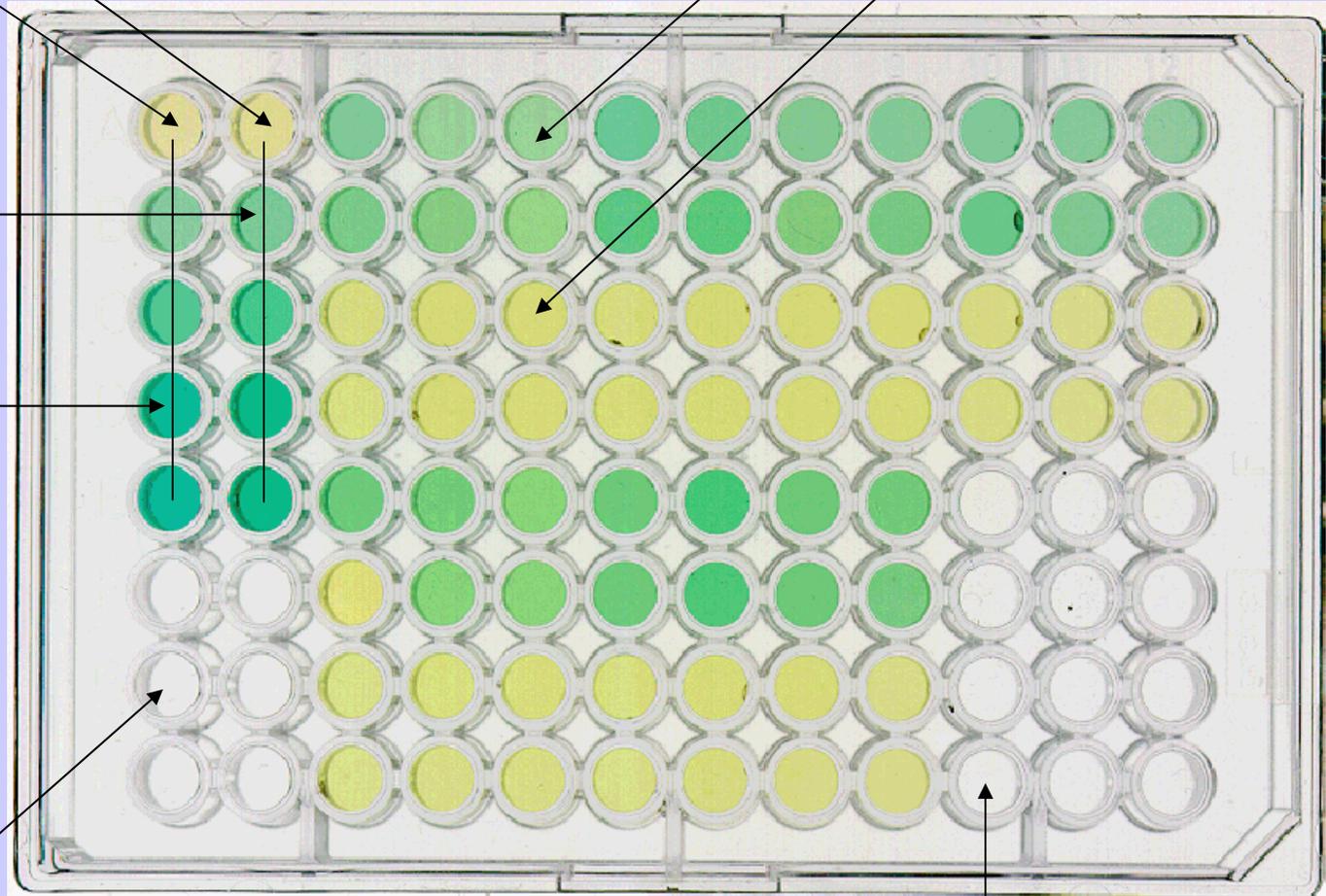


# Mikroplattenphotometer

Reagenzienleerwerte

Vollproben  
Blindproben

Kalibrationsreihen



Reines Lösungsmittel (A.demin.)

$V = \text{konstant!}$

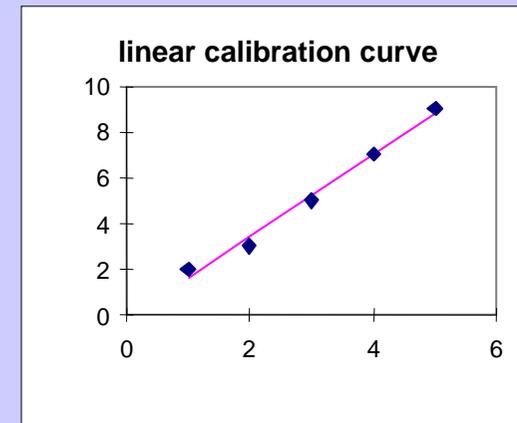
# Auswertung (Rechnung) über Kalibrationskurve

Position	Sample ID	Abs	Conc
A 1	Kalibrationspunkt 1	1	2
B 1	Kalibrationspunkt 2	2	3
C 1	Kalibrationspunkt 3	3	5
D 1	Kalibrationspunkt 4	4	7
E 1	Kalibrationspunkt 5	5	9
F 1			
G 1			
H 1			

$y' = kx + d$   
 $k = 1,80$   
 $d = -0,2$   
 $r = 0,9938837$

Factor	Abs	Conc
1	1	1,6

Copy this formula to column D to calculate the  $y'$  values!



$y'$   
 1,6  
 3,4  
 5,2  
 7  
 8,8  
 -0,2  
 -0,2  
 -0,2

Wiederholung

# Kalibrationskurve

